

Mars-surface minerals in the hyper-acid lake of Poás volcano, Costa Rica

Alejandro Rodríguez¹, Manfred van Bergen¹, Annika Huizinga¹, Georgina Ayres¹,
María Martínez² and Erick Fernández²

¹Department of Earth Sciences, Utrecht University, Budapestlaan 4, 3584 CD Utrecht, The Netherlands
²Observatorio Vulcanológico y Sismológico de Costa Rica Universidad Nacional (OVSICORI-UNA), P.O. Box 86-3000 Heredia, Costa Rica
Contact e-mail: a.rodruiguezbadilla@uu.nl

1. INTRODUCTION

Orbital and landed missions have provided evidence for the widespread occurrence of sulfate-rich mineral associations across the Martian landscape (Swayse *et al.*, 2008; Ehlmann *et al.*, 2011; among others). They must have formed under acidic and oxidizing conditions in the presence of water (Xu *et al.*, 2010). Active volcanoes with crater lakes on Earth are SO₄⁻ and Cl⁻ dominated systems in which sulfate-forming processes can be studied *in situ*. For this purpose, we are exploring the active crater of Poás volcano (Costa Rica) that hosts Laguna Caliente, a hot (T>45°C) and hyperacid (pH<2.0) sulfate-chloride lake (Figs. 1 and 2).

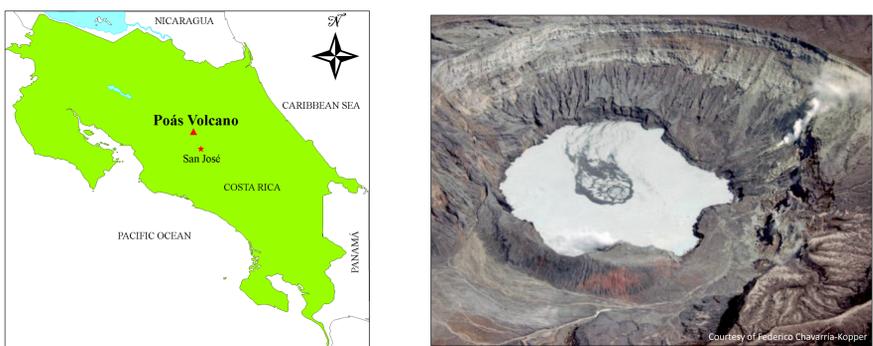


Fig. 1. Location map and aerial view of Laguna Caliente

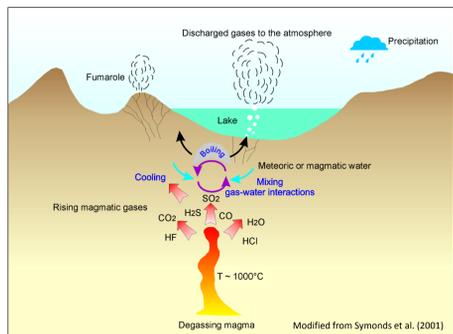


Fig. 2. Conceptual model of a volcanic lake

2. SECONDARY MINERALOGY

SEM, EPMA and XRD techniques revealed the presence of alunite (KAl₃(SO₄)₂(OH)₆) in a clastic sedimentary rock found in the lake shore. It presumably originated in the bottom of Laguna Caliente and must have been expelled during a recent explosive phreatic event. Additionally, anhydrite, amorphous silica, cristobalite, rhomboclase ((H₅O₂)Fe(SO₄)₂•2H₂O) and bilinite (Fe²⁺Fe³⁺₂(SO₄)₂•22H₂O) were identified by powder XRD in a bottom sediment from Laguna Caliente (Figs. 3 and 4).

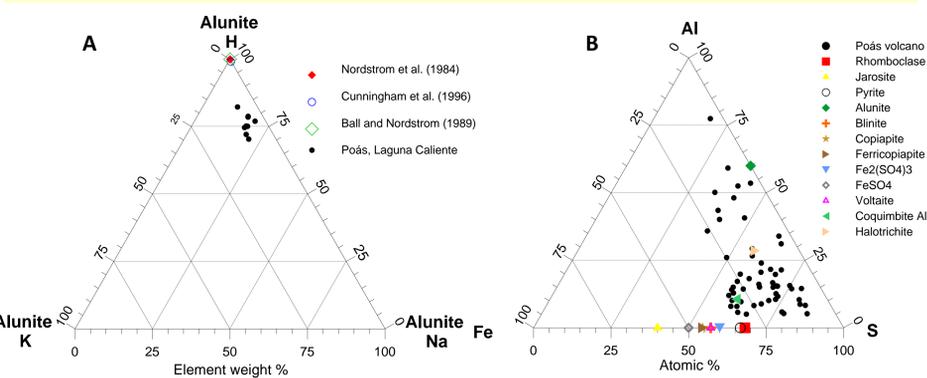


Fig. 3. EPMA analyses of a clastic sedimentary rock (A) and bottom lake sediment (B)

6. REFERENCES

Cigolini, C., Kudo, A.M., Brookins, D.G. and Ward, D., 1991. The petrology of Poás volcano lavas: basalt-andesite relationship and their petrogenesis within the magmatic arc of Costa Rica. *Journal of Volcanology and Geothermal Research*, 48(3-4): 367-384.
Ehlmann, B.L., Mustard, J.F., Murchie, S.L., Bibring, J.-P., Meunier, A., Fraeman, A.A. and Langevin, Y., 2011. Subsurface water and clay mineral formation during the early history of Mars. *Nature*, 479(7371): 53-60.
Martínez, M., 2008. Geochemical evolution of the acid crater lake of Poás volcano (Costa Rica): Insights into volcanic-hydrothermal processes. PhD Thesis, Utrecht University, 162 pp.
Parkhurst, D.L. and Appelo, C.A.J., 1999. User's guide to PHREEQC (Version 2)—a computer program for speciation, batch-reaction, one-dimensional transport, and inverse geochemical calculations. Report 99-4259, U.S. Geological Survey, Water-Resources Investigations.
Swayze, G.A., Ehlmann, B.L., Milliken, R.E., Poulet, F., Wray, J.J., Rye, R.O., Clark, R.N., Desborough, G.A., Crowley, J.K., Gondet, B., Mustard, J.F., Seelos, K.D. and Murchie, S.L., 2008. Discovery of the Acid-Sulfate Mineral Alunite in Terra Sirenum, Mars, Using MRO CRISM: Possible Evidence for Acid-Saline Lacustrine Deposits?. *American Geophysical Union, Fall Meeting 2008*, abstract #P44A-04.
Symonds, R.B., Gerlach, T.M. and Reed, M.H., 2001. Magmatic gas scrubbing: implications for volcano monitoring. *Journal of Volcanology and Geothermal Research*, 108: 303-341.
Xu, W., Parise, J.B. and Hanson, J., 2010. (H₃O)Fe(SO₄)₂ formed by dehydrating rhomboclase and its potential existence on Mars. *American Mineralogist*, 95(10): 1408-1412.

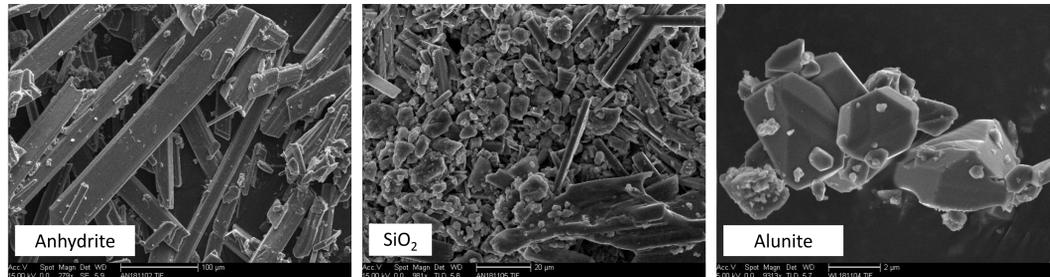


Fig. 4. SEM images of some secondary minerals

3. SATURATION INDICES

Speciation calculations carried out with PHREEQC (Parkhurst and Appelo, 1999) on data from Laguna Caliente (Martínez, 2008) show that lake waters are in equilibrium with amorphous silica, anhydrite, cristobalite, sulfur, pyrite and are undersaturated in alunite, jarosite, fluorite and epsomite. Increasing temperatures lead to kaolinite, AlF₃, and eventually alunite supersaturation whereas pyrite and sulfur become undersaturated (Fig. 5).

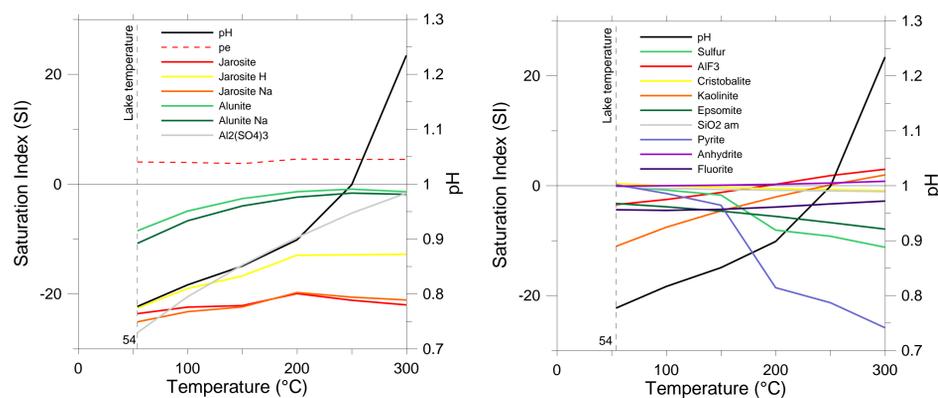


Fig. 5. Variation of mineral saturation indices with temperature for a sample taken on November 30th, 2005 in Laguna Caliente

4. REACTION PATH MODELING

Reaction path modeling was carried out in PHREEQC (Parkhurst and Appelo, 1999). Anatase, amorphous silica, nontronite-H and anhydrite form at low reaction progress whereas alunite, kaolinite, and eventually pyrite and fluorite appear with increased rock dissolution (Fig. 6).

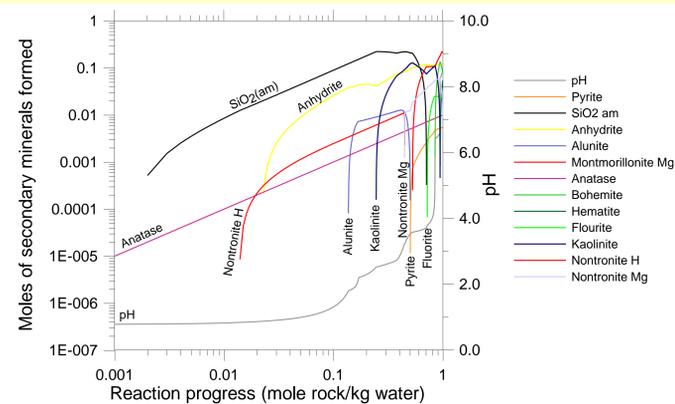


Fig. 6. Reaction path modeling of 110 g of a basaltic andesite from Poás volcano. Rock analysis reported by Cigolini et al. (1991)

5. ONGOING WORK

Field work will be carried out during April 2012 in order to collect water and gas samples from Laguna Caliente. Additionally, the distribution of alteration minerals around the crater will be studied. Sulfur, chlorine and bromine will receive special attention as their concentrations and isotopes will provide valuable information on pathways and formation mechanisms of sulfates and halides.

This work is financially supported by the NWO Division for the Earth and Life Sciences (ALW), cooperating with the Netherlands Space Office (NSO) in the User Support Space Research programme.