

# Species specific impacts of temperature and seawater Mg/Ca on foraminiferal Mg/Ca

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## Introduction

Variations in Mg/Ca of foraminiferal calcite ( $Mg/Ca_{\text{calcite}}$ ) reflect changes in environmental parameters, particularly temperature and seawater Mg/Ca ( $Mg/Ca_{\text{sw}}$ ) (Fig. 1). This means that  $Mg/Ca_{\text{calcite}}$  can only be used to reconstruct seawater temperature when  $Mg/Ca_{\text{sw}}$  is known.

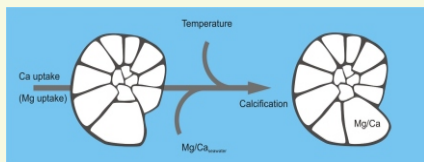


Figure 1. Mg incorporation in a foraminifera, influenced by  $Mg/Ca_{\text{sw}}$  and temperature.

The impact of  $Mg/Ca_{\text{sw}}$  is negligible when reconstructing paleo-temperatures over timescales within  $\sim 1$  Ma, due to the long oceanic residence times of  $Mg^{2+}$  and  $Ca^{2+}$ . However, when reconstructing seawater temperatures on longer timescales, T reconstructions have to be corrected for  $Mg/Ca_{\text{sw}}$  (Fig. 2).

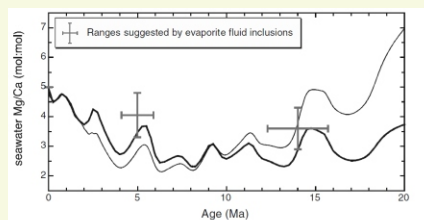


Figure 2. The seawater Mg/Ca ratio over the past 20 million years, as suggested by evaporite (crosses) and carbonate sediment (curves) fluid inclusions (Fantle and DePaolo, 2006).

Incorporation of Mg into foraminiferal calcite ( $Mg/Ca_{\text{calcite}}$ ) is expressed as the partition coefficient:

$$D_{\text{Mg}} = (Mg/Ca_{\text{calcite}})/(Mg/Ca_{\text{sw}})$$

$D_{\text{Mg}}$  in turn may vary with temperature and preliminary data shows that  $D_{\text{Mg}}$  is also impacted by  $Mg/Ca_{\text{sw}}$ .

## Objectives

The use of low- and Mg-calcite species from the same age allows reconstruction of both  $Mg/Ca_{\text{sw}}$  and temperature. To do this, it is necessary to investigate the dependency of temperature and  $Mg/Ca_{\text{sw}}$  on  $D_{\text{Mg}}$ .

## Conclusions

Preliminary results suggest that for *Quinqueloculina* sp.,  $Mg/Ca_{\text{sw}}$  and T impact  $D_{\text{Mg}}$  independently.

## Further research suggestions

Further research can use this relationship in combination with a species with a contrasting biologically controlled Mg incorporation (e.g., the low Mg-producer *Elphidium* sp.) to resolve  $Mg/Ca_{\text{sw}}$  changes in deep geological time, since the relative difference reflects temperature and  $Mg/Ca_{\text{sw}}$ . This approach takes advantage of the species-specific fractionation between  $Mg^{2+}$  and  $Ca^{2+}$ .

## Methods

Specimens of *Quinqueloculina* sp. were sampled in Tokyo bay, Japan and were cultured at a combination of different temperatures (17, 22 and 27 °C) and  $Mg/Ca_{\text{sw}}$  (2, 3.5 and 5 mol/mol).  $Mg/Ca_{\text{calcite}}$  was determined using laser ablation-inductively coupled plasma-mass spectrometry (LA-ICP-MS; Reichart et al., 2003) and expressed as elemental concentrations (Fig. 3).

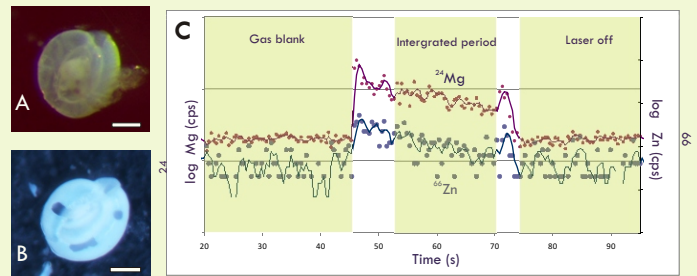


Figure 3. *Quinqueloculina* sp. (scalebar = 100  $\mu$ m) during culturing (a) and after cleaning and laser ablation (b). Results from laser ablation (c) with  $^{24}\text{Mg}$  and  $^{66}\text{Zn}$  in cps and indication of the integrated period.

## Preliminary results

With increasing  $Mg/Ca_{\text{sw}}$  as well as with increasing temperature,  $Mg/Ca_{\text{calcite}}$  increases (Fig. 4). The different temperature relations appear to have a similar slope. Although it is suggested that  $D_{\text{Mg}}$  may vary with temperature, this appears to be not the case for this species.

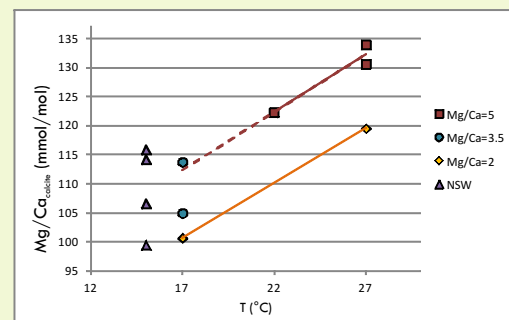
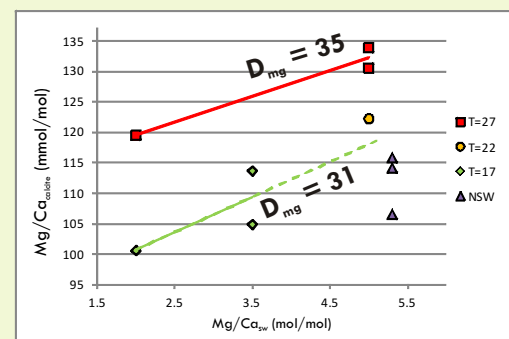


Figure 4.  $Mg/Ca_{\text{calcite}}$  of cultured *Quinqueloculina* sp. Upper:  $Mg/Ca_{\text{calcite}}$  vs.  $Mg/Ca_{\text{sw}}$  with  $D_{\text{Mg}}$ . Lower:  $Mg/Ca_{\text{calcite}}$  vs. temperature. 'NSW' represents chambers grown at natural seawater conditions, and reflect  $Mg/Ca_{\text{sw}} \approx 5.3$  and  $T \approx 15^\circ\text{C}$ .



## References

- Reichart, G.J., Jorissen, F., Anschutz, P., Mason, P.R.D. (2003) Single foraminiferal test chemistry records the marine environment. *Geology* 31, 355-358.
- Fantle, M.S., and DePaolo, D.J. (2006) Sr isotopes and pore fluid chemistry in carbonate sediment of the Ontong Java Plateau: Calcite recrystallization rates and evidence for rapid rise in seawater Mg over the last 10 million years. *Geochimica et Cosmochimica Acta* 70, 3883-3904.