# Process driven engineering of sorbents for optimal CO<sub>2</sub> capture from air



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### **1** Introduction

- There exist hundreds of thousands of solid sorbent materials, both real and hypothetical, mainly developed to capture CO<sub>2</sub> from point sources.
- In this study, we first scan existing materials to assess their suitability to capture CO<sub>2</sub> from diluted sources, e.g. direct air capture (DAC).
- a second step, we carry out a theoretical exercise and optimize the • In parameters of different isotherm models, representing different isotherm shapes.
- By doing so, we can investigate the limits of CO<sub>2</sub> adsorption from dilute feed

## 2 Modeling tools

#### Methodology

- Simplified 0D model of a four-step ● temperature vacuum swing (TVSA) adsorption system.
- The model is based on a batch ulletsystem and each step is modeled adsorption assuming by equilibrium.



streams, in terms of process performance as well as costs.

• Furthermore, we can compare these optimal isotherms with existing materials.

## **3** Sorbent screening

Screening of the NIST/ARPA-E database (around 2500 materials) and 80  $\bullet$ additional adsorbents added by hand<sup>2-4</sup>.



- Varying the  $CO_2$  concentration in the feed (0.04%, 0.1%, 1.0%).
- For y<sub>co2</sub>>0.04%, the capture rate is constrained to >90%.





- Extended by incorporating a neural network, trained with a rate-based model, to include productivity as a performance parameter, besides the specific thermal energy consumption, the purity as well as the recovery.
- In addition, we consider saturation levels in the bed during adsorption below 100%.
- Considering a ternary mixture as feed, where CO<sub>2</sub> is not necessarily the most retained gas, i.e. we add H<sub>2</sub>O adsorption.
- Including multiple CO<sub>2</sub> isotherm types in the model, i.e. Toth, extended Toth ulletmodel (Toth-cp), Langmuir-Freundlich, Dual-Site-Langmuir (DSL) and sshaped isotherm model.

#### Model validation

- Validation using a rate-based model<sup>1</sup>
- Two steps:  $\bullet$

![](_page_0_Figure_32.jpeg)

#### **Process-optimization**

![](_page_0_Figure_35.jpeg)

#### **4** Sorbent isotherm optimization

![](_page_0_Figure_37.jpeg)

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